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introduced in the chapter and the skills section provides developed exercises to apply the new concepts in the chapter to particular examples the common mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the practical test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter after briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding these topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid the remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry this study guide will be of great value to chemistry teachers and students

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book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter s main points to aid in reader comprehension important terms are defined in context and key ideas are summarized many thought provoking discussion questions worked examples and end of chapter problems are also included each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter in addition each chapter has an historical note exploring connections with the people and cultures connected to topics in the text a problem solving approach to aquatic chemistry includes fundamental concepts such as concentration units thermodynamic basis of equilibrium and manipulating equilibria solutions of chemical equilibrium problems including setting up the problems and algebraic graphical and computer solution techniques acid base equilibria including the concepts of acids and bases titrations and alkalinity and acidity complexation including metals ligands equilibrium calculations with complexes and applications of complexation chemistry oxidation reduction equilibria including equilibrium calculations graphical approaches and applications gas liquid and solid liquid equilibrium with expanded coverage of the effects of global climate change other topics including chemical kinetics of aquatic systems surface chemistry and integrative case studies for advanced senior undergraduates and first year graduate students in environmental engineering courses a problem solving approach to aquatic chemistry serves as an invaluable learning resource on the topic with a variety of helpful learning elements included throughout to ensure information retention and the ability to apply covered concepts in practical settings

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