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write the formula of the salt formed when a given acid and base are combined the concepts of an acid a base and a salt are very old ones that have undergone several major refinements as chemical science has evolved

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we can determine whether a salt solution will be acidic basic or neutral by considering the reactivity of both the cation and the anion with water if neither species reacts with water the solution will be neutral

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what is salt in acids bases and salts in chemistry a salt is a substance obtained by the reaction of an acid and a base salts are composed of positive ions cations of bases and negative ions anions of acids

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write chemical reactions between an arrhenius acid and an arrhenius base and between brønsted lowry acid and a brønsted lowry base there are three major classifications of substances known as acids or bases

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ph and solubility solubility and complex ion formation this unit is part of the chemistry library browse videos articles and exercises by topic

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in this unit we will be talking about the basic concepts of acids bases and salts we will cover acid and base definitions reactions strength ph chemical compounds of salts and their uses

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salts can be neutral acidic or basic because they contain the conjugate acids and bases of the bases and acids that formed them knowing the ionization constant of an acid or base allows you to

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acids bases and salts ph scale and indicators many chemicals can be classified as being acidic neutral or alkaline by using indicators the ph scale is used to measure acidity and

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base chemistry appearance hide for other uses see base disambiguation see also alkali soaps are weak bases formed by the reaction of fatty acids with sodium hydroxide or potassium hydroxide acids and bases acceptor number acid base reaction acid base homeostasis acid strength acidity function amphoterism base buffer solutions

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the sodium hydroxide calcium carbonate and potassium oxide are examples of bases a base is a substance that reacts with hydrogen ions and can neutralize the acid most bases are minerals which form water and salts by reacting with acids bases include the metal oxides hydroxides and carbonates

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there are three different definitions of acids and bases in this section we will introduce the fundamentals of acid base chemistry acids taste sour make metals corrode change litmus a dye from plants red and get neutralized when bases are added

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learn the basics about what salts are as part of the overall topic of acids and bases subscribe to the fuse school youtube channel for many more educational videos

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bases are a group of substances that neutralise acids soluble bases are called alkalis they have a slippery soapy feel and form solutions with ph values greater than 7 common examples include sodium hydroxide magnesium hydroxide sodium hydroxid

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a substance is called base if its aqueous solution tastes bitter turns red litmus blue or neutralizes acids salt is a neutral substance whose aqueous solution does not affect litmus according to faraday acids bases and salts are termed as electrolytes

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this chemistry video tutorial provides a basic introduction into acids and bases it explains how to identify acids and bases in addition to how they react with water it discusses how to

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acids bases and salts propertries of acids and bases general properties of acids and bases what are acids and bases acids are chemical substances which are characterized by a sour taste in an aqueous medium they have the tendency to turn blue litmus red

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why does a salt containing a cation from a weak base and an anion from a strong acid form an acidic solution how do the k a or k b values help determine whether a weak acid or weak base will be the dominant driving force of a reaction

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identify salts as neutral acidic or basic khan academy

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see salts they can be both acidic and basic as well in this video we are going to take some salts and try to identify their nature so first of all lets begin with what we already know so we know that acids and bases when they react they neutralize each other s effect

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