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a simple explanation of what alloys are and how they re made with a table listing some of the more common alloys

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in an alloy the atoms are joined by metallic bonding rather than by covalent bonds typically found in chemical compounds the alloy constituents are usually measured by mass percentage for practical applications and in atomic fraction for basic science studies

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an alloy is a material composed of a metallic base and additional metal or non metal elements alloys are typically created to obtain desirable material properties that otherwise are not present in the primary base material s pure form

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alloy metallic substance composed of two or more elements as either a compound or a solution the components of alloys are ordinarily themselves metals though carbon a nonmetal is an essential constituent of steel alloys are usually produced by melting the mixture of ingredients

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an alloy is a substance made by combining together two or more elements where the primary element is a metal most alloys form by melting the elements together upon cooling an alloy crystallizes into a solid intermetallic compound or mixture that cannot be separated using a physical method

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alloys are metals made of more than one element that s it as long as the final product behaves like a metal but it s not elementally pure it s an alloy there are no special rules about solid solution number of phases or the ratio of metallic and non metallic elements

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