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thermodynamics of solutions a useful quantity to characterize the solution composition is the so called mole fraction defined as follows suppose a solution contains n components let there be n_1 moles of component 1 n_2 moles of component 2 \dots n_n moles of component n

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solution since an vrt is a power it is dimensionless and a has the same dimensions as vrt n these dimensions are volume energy amount² expressed in m³ jmol² b has the same dimensions as v n which are volume amount expressed in m³ mol¹ b the redlich kwong equation p d r t v n b an² t¹ 2v v cnb solution

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