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learning objectives to understand how enthalpy and entropy changes affect solution formation to use the magnitude of the changes in both enthalpy and entropy to predict whether a given solute solvent combination will spontaneously form a solution

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physical and chemical processes can be classified by the changes occurring on the molecular level in general chemical processes involve changes in chemical bonds while physical processes involve changes only in intermolecular forces

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in chemistry a solution is a special type of homogeneous mixture composed of two or more substances in such a mixture a solute is a substance dissolved in another substance known as a solvent

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